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▶ To cite this version:

Melody Briard, Clément Brandel, Gérard Coquerel, Valerie Dupray. On the mechanism of Non-Photochemical Laser Induced Nucleation. CRYSTAL GROWTH OF ORGANIC MATERIALS - CGOM14, Sep 2022, Brussels, Belgium. hal-03777825

HAL Id: hal-03777825 https://normandie-univ.hal.science/hal-03777825

Submitted on 15 Sep 2022

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ON THE MECHANISM OF NON-PHOTOCHEMICAL LASER INDUCED NUCLEATION





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NPLIN = Non-Photochemical Laser Induced Nucleation

Nucleation process discovered by Garetz et al¹ (1996)

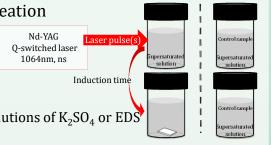
NPLIN allows: - A temporal control of nucleation (\sinduction time)

- An increase of the probability of nucleation
- A control of the number of crystals

What is the mechanism of this phenomenon?

Different hypotheses have been investigated through supersaturated aqueous solutions of K₂SO₄ or EDS

Cavitation, Optical Kerr Effect and nanoparticle heating



Cavitation: bubble generation²

Hypothesis:

irradiation

nucleation

collapse

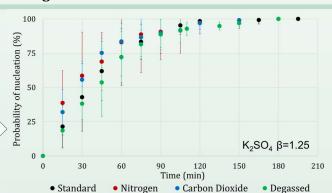
Crystal(s) formation

If bubbles were composed of dissolved gas

→ Different gas should lead to different NPLIN kinetics

<u>Experiment</u>: Bubbling of gas to control composition and nature Results of gases in supersaturated aqueous solutions of K₂SO₄

> → NPLIN behavior is similar for each gas condition, There is no influence of gases



Can NPLIN be a stereoselective process?

<u>Hvpothesis</u>: OKE (Optical Kerr Effect¹)

If an alignment of the dipoles along the electromagnetic field occurs

Then, laser polarization can influence NPLIN



Results: EDS β=1.10

100

Polarization	N_{vial}	N _{crystal}	N _D	N _L	Cee±σ
Left Circularly Polarized Light	89	194	99	95	0.02±0.07
Right Circularly Polarized Light	86	190	81	109	-0.14±0.07
Linearly Polarized Light	91	251	129	122	0.03±0.06

→ Crystal Enantiomeric Excess (Cee) shows no significant change upon changing laser polarization NPLIN is non-stereoselective. OKE is not applicable to EDS.

Nanoparticle heating ³

Hypothesis: Solid particles play a role in the mechanism



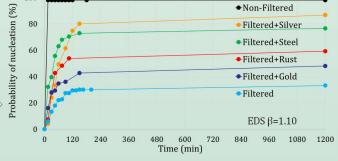


- a) Absorption of laser light b) Cavitation (vapor bubble)
- c) Pressure waves d) Destruction of the particle

Experiment: Filtration and doping of supersaturated solution Results ICP analysis of removed impurities

Concentration (mg/L) Control 0.28 0.06 0.14 **Impurity Sample** 8.2 0.76 0.71 1.60

→ Stainless steel Composition



→ Filtration decreases the probability of nucleation, but addition of different particle lead to different kinetics behavior

Conclusions

Laser polarization geometry has no incidence on nucleation:

U→**OKE mechanism** is invalid for EDS

If there is **Cavitation**, bubbles might not be composed of gas

→ 2nd hypothesis bubbles might be composed of solvent vapor

Nanoparticle heating seems the best hypothesis since presence or absence of particles has a strong influence on NPLIN behavior but it is not clear how particles are involved in the nucleation of the crystal (Hypothesis b, c or d from(4)).

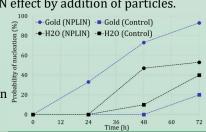
Perspectives

Possibility to enhance the NPLIN effect by addition of particles.

For Diprophylline/ H_2O β =3.25 Doping with Gold nanoparticles

- Induction time
- → Probability of nucleation

Different particles concentration and nature have to be tested



(1) Garetz, B. A.; Aber, J. E.; Goddard, N. L.; Young, R. G.; Myerson, A. S. Physical review letters 1996, 77 (16), 3475 (2) Soare, A.; Dijkink, R.; Pascual, M. R.; Sun, C.; Cains, P. W.; Lohse, Stankiewicz, A. I.; Kramer, H. J. M. Crystal Growth & Design 2011, 11 (6), 2311–2316. (3) Ward, M. R.; Mackenzie, A. M.; Alexander, A. J. Crystal Growth & Design 2016, 16 (12), 6790–6796. Alexander, A. J.; Camp, P. J.. J. Chem. Phys. 2019, 150 (4), 040901