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Chiral purity limitation: discovery of a partial solid solution

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SEOZNS

Context

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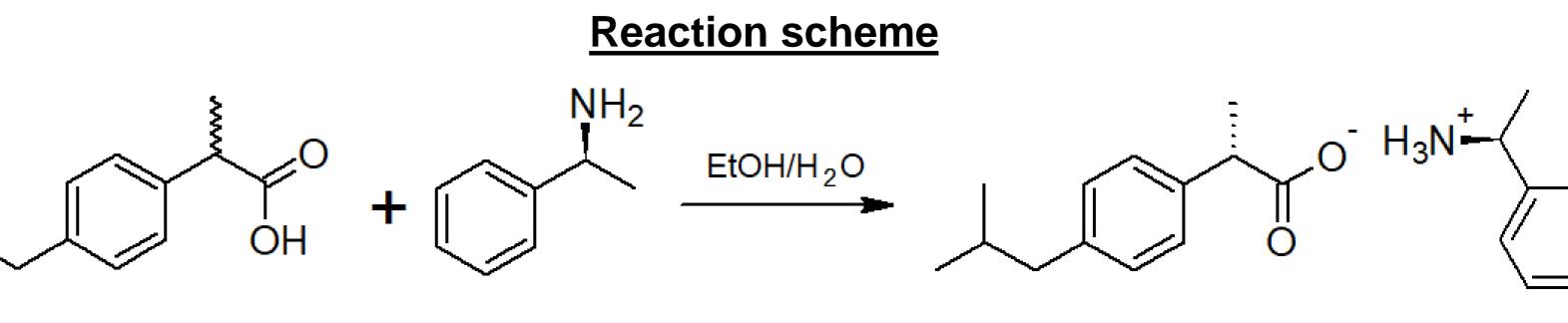


Normandie Université

Chiralité et Multifonctionnalités

Greater therapeutic effect for S-Ibuprofen (S-Ibu) [1]

- \forall Chiral resolution by crystallization using S- α -
 - MethylBenzylAmine (**S-αMBA**)
 - Sormation of S-Ibuprofen-S-α-MethylBenzylAmine (S-lbu-S-αMBA)



Racemic Ibuprofen

S-α-MethylBenzylAmine

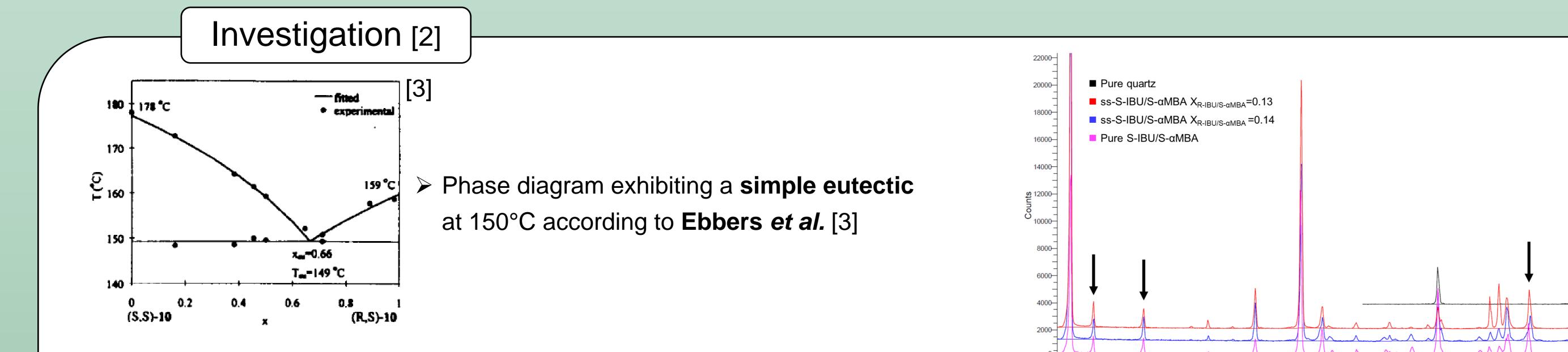
S-Ibuprofen-S-α-MethylBenzylAmine

 \Rightarrow Observed limit of the chiral purity reachable (87% of S-lbu-S- α MBA)

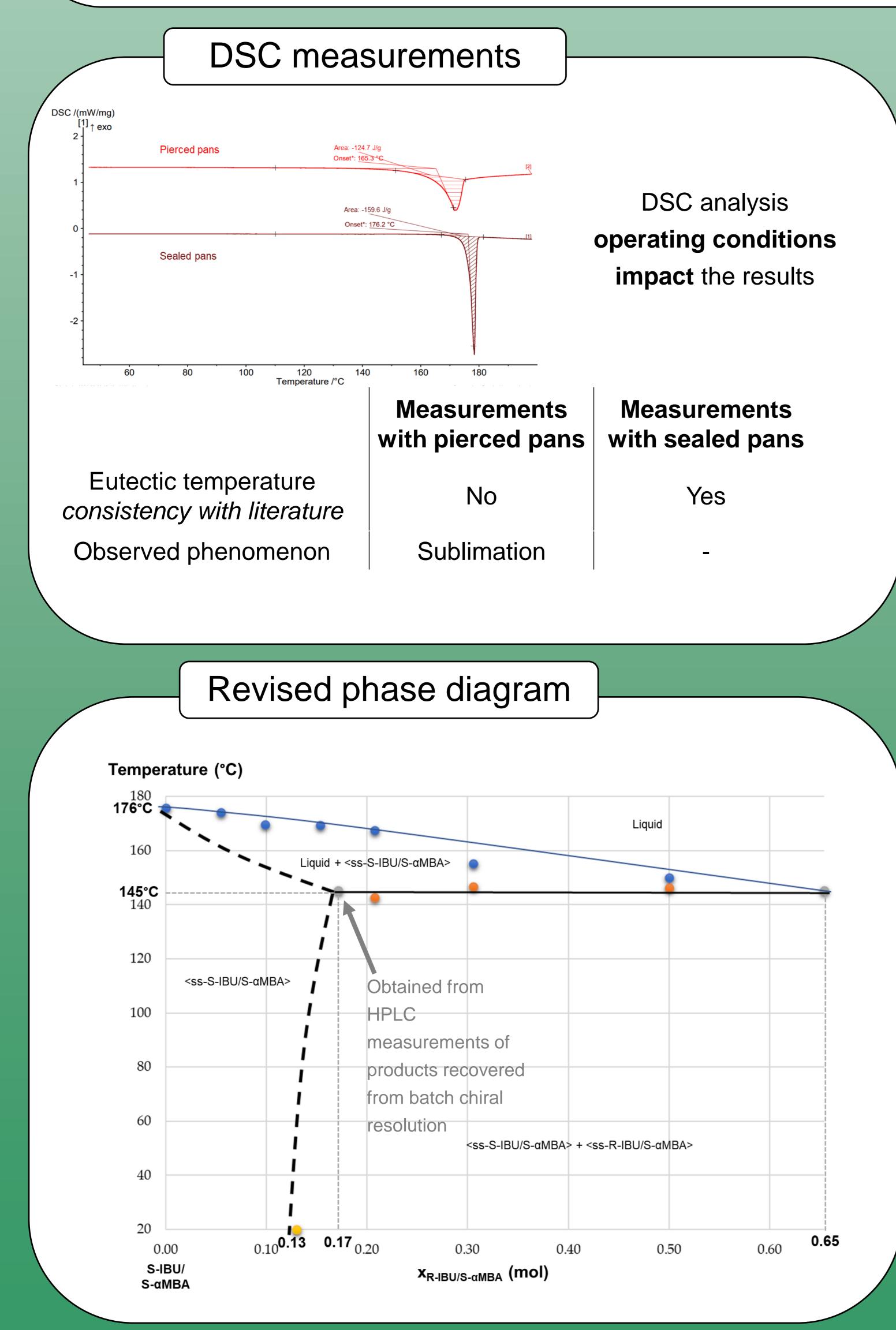
Existence of a partial solid solution restraining the maximum chiral purity

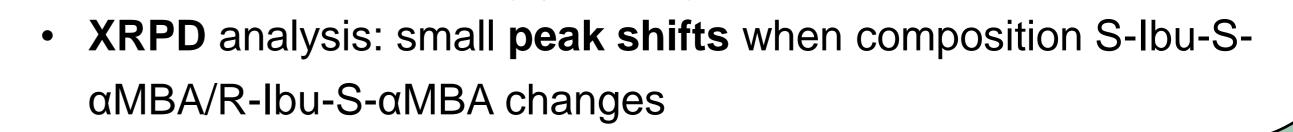
\Rightarrow Detection of **peak shifts** in **XRPD** analysis

accessible?

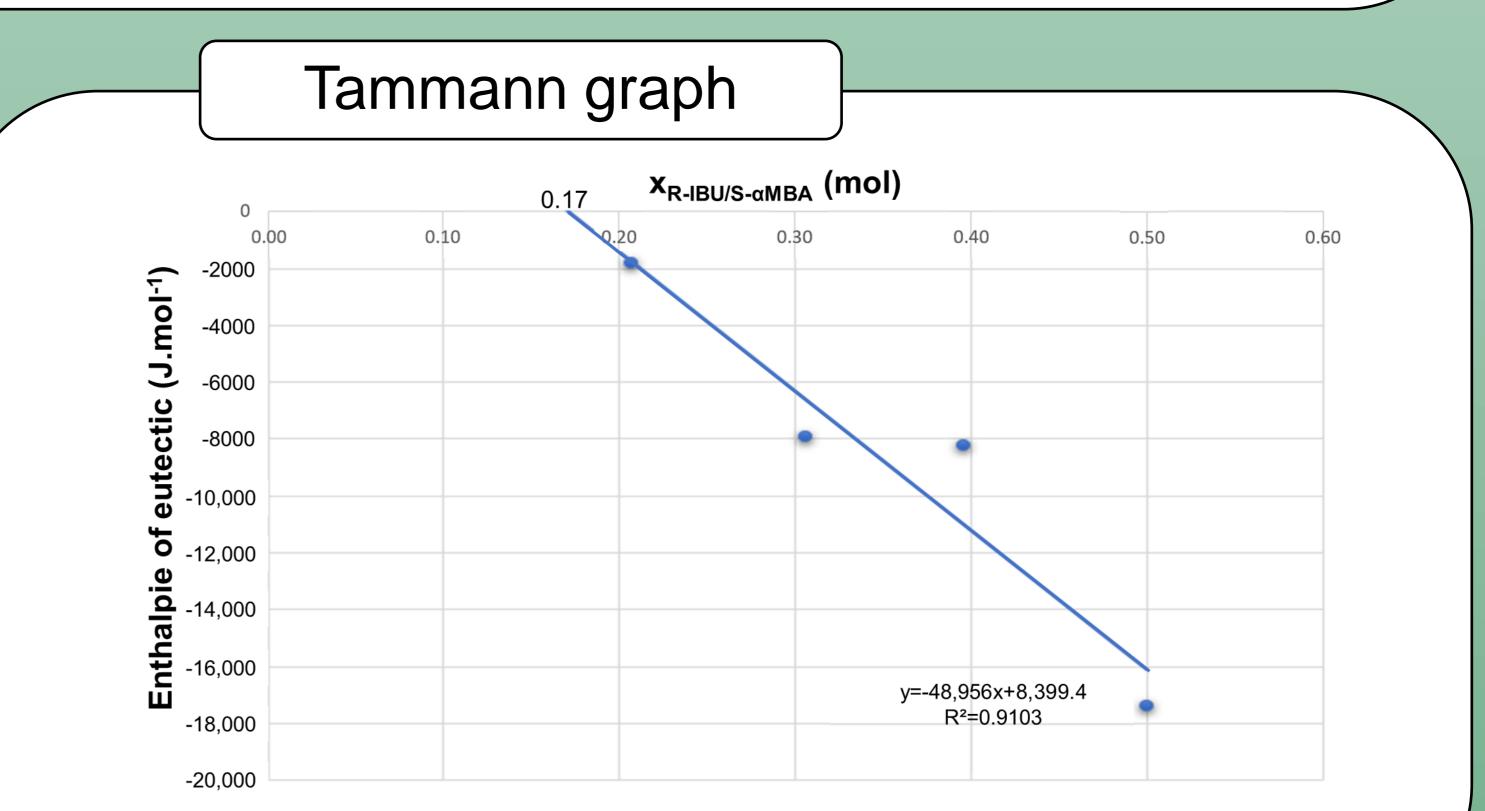


- **DSC measurements** at **different compositions** between X_{R-Ibu-S-αMBA}=0 and 0.5 \bullet \clubsuit Melting temperatures and eutectic invariant temperature \rightarrow Phase diagram
 - \clubsuit Eutectic enthalpies \rightarrow Tammann graph





2Theta (Coupled TwoTheta/Theta) WL=1,54060



The intersection with abscissa axis gives the eutectic composition.

Conclusions

- The **Schröder van Laar** equation has been used to confirm the eutectic composition and temperature. Calculated results are **consistent** with the experimental ones.
- Contrary to published data [3], the **binary system between S-Ibu-SαMBA/R-Ibu-S-αMBA** does not only exhibit a simple eutectic:

- A partial solid solution has been evidenced for the less soluble salt [2].
- This partial solid solution is the **cause** of the **limitations in chiral** purity.
- In order to prevent sublimation, **DSC** measurements were performed in sealed pans. In sealed pans, pressure increases with temperature, which induces a measurement bias:
 - Solution The 'real' phase diagram cannot be determined with usual DSC apparatus.

[References]. [1] A. M. Evans, Clinical Rheumatology, vol20, n°1, p. 9-14 (2001). [2] L. Marc *et al.*, Tetrahedron: Asymmetry, **vol. 8**, n°24, p. 4047-4057. (1997).