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Continuous chiral resolution by diastereomeric salt formation of racemic Ibuprofen in a Couette-Taylor crystallizer

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Batch production mode [1]

- legacy in pharmaceutical industry

Introduction

- weaknesses such as batch-to-batch quality variation

Continuous production mode [2]

- steady-state functioning, i.e. more constant quality product
- better process control

35.0

25.0

20.0

15.0

5.0

Reaction scheme

 NH_2





Racemic Ibuprofen L-α-MethylBenzylAmine

S-Ibuprofen-L-α-MethylBenzylAmine

T1: TEMPERATURE OF THE EXTERNAL CYLINDER

T2: TEMPERATURE OF THE INTERNAL CYLINDER



TAYLOR VORTICES

Batch	n process 🛏				
Conditions Scale: 10g		Dg Stirring speed (m 200	Stirring speed (mechanical stirrer): 200rpm		
Results	Yield	Diastereomeric excess	Diastereomeric yield		
	41%	0.58	24%	Vac-High PC-Std. 10 kV x 100 200 µm 017445	Vac-High PC-Std. 10 kV x 1000 20 μm 017447
Cont	inuous proc	NUOUS ATION	Rac-lbu IL FEED TANKS	Condition Scale: Stirring spectrum 70g 1000rpr	DNS ed Ω: Flowrate: n 3.34mL/min



=> Variation of the temperature difference between the two cylinders with T2>T1

$\Delta T=20^{\circ}C$







quickly

- - \Rightarrow Crystal size increases when the

temperature difference increases

 \Rightarrow At this ΔT , the product quality variation can be due to the crystal size

 $\Rightarrow \Delta T=10^{\circ}C$ seems to give the best results regarding product quality

Conclusions and perspectives

- > The yield of the continuous process is close to that obtained in batch mode. By contrast, the chiral purity is lower than that of the batch mode. The reason might be linked to the **filtration process** which is not adapted to the continuous mode.
- \Rightarrow It is noteworthy that the **combination of both \Omega and \Delta T** can have an influence on chiral purity and crystal size [5].
- \Rightarrow The continuous process needs to be optimized by studying variations of the stirring speed Ω , the flowrate, the temperature inside the CT crystallizer, and the sign of ΔT .

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