

# Phytochemical constituents from the seeds of Calycotome villosa subsp. intermedia

Aouicha Elkhamlichi, Ali El Antri, Hanane El Hajaji, Brahim El Bali, Hassan

Oulyadi, Mohammed Lachkar

### ▶ To cite this version:

Aouicha Elkhamlichi, Ali El Antri, Hanane El Hajaji, Brahim El Bali, Hassan Oulyadi, et al.. Phytochemical constituents from the seeds of Calycotome villosa subsp. intermedia. Arabian Journal of Chemistry, 2017, 10 (S2), pp.S3580-S3583. 10.1016/j.arabjc.2014.03.005 . hal-02046274

## HAL Id: hal-02046274 https://normandie-univ.hal.science/hal-02046274

Submitted on 22 May 2024  $\,$ 

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - NonCommercial - NoDerivatives 4.0 International License



### King Saud University

### Arabian Journal of Chemistry

www.ksu.edu.sa



www.sciencedirect.com

### **ORIGINAL ARTICLE**

# Phytochemical constituents from the seeds of Calycotome villosa subsp. intermedia



### Aouicha Elkhamlichi<sup>a</sup>, Ali El Antri<sup>a</sup>, Hanane El Hajaji<sup>a</sup>, Brahim El Bali<sup>b</sup>, Hassan Oulyadi<sup>c</sup>, Mohammed Lachkar<sup>a,\*</sup>

<sup>a</sup> Engineering Laboratory of Organometallic and Molecular Materials "LIMOM". CNRST (URAC 19). Faculty of Sciences Dhar El Mahraz, University Sidi Mohamed Ben Abdellah, P.O. Box 1796 (Atlas), 30000 Fez, Morocco <sup>b</sup> Laboratory of Mineral Solid and Analytical Chemistry 'LMSAC', Department of Chemistry, Faculty of Sciences, University Mohamed I, P.O. Box 717, 60000 Oujda, Morocco

<sup>c</sup> UMR 6014 COBRA & FR3030 INC3M, Equipe Analyse et Modélisation, Université de Rouen – IRCOF, Rue Tesnière, 76821 Mont Saint-Aignan, France

Received 26 July 2013; accepted 5 March 2014 Available online 13 March 2014

#### **KEYWORDS**

Calycotome villosa subsp. intermedia; Seeds: Dihydroisoquinoline-N-oxide: Methyl 4-hydroxybenzoate; X-ray crystal structure determination

Abstract A new dihydroisoquinoline-N-oxide alkaloid 1 together with the previously known methyl 4-hydroxybenzoate 2 were isolated for the first time from the seeds of *Calvcotome villosa* subsp. intermedia. Their structures were elucidated by extensive spectroscopic analysis and in the case of the new alkaloid 1-hydroxymethyl-6,7-dimethoxy-3,4-dihydroisoquinoline-N-oxide 1 by X-ray diffraction analysis.

© 2014 King Saud University. Production and hosting by Elsevier B.V. This is an open access article under the CC BY-NC-ND license (http://creativecommons.org/licenses/by-nc-nd/3.0/).

#### 1. Introduction

Calycotome villosa (Poiret) Link (Leguminosae), a shrub that grows mostly in cool places, is very common in the Mediterranean area (Gibbs, 1968; Tutin, 1972). However, C. villosa (Poiret) Link subsp. intermedia (C. Presl) Quezel and Santa, a 50-150 cm spiny shrub with yellow flowers during the spring season, grow

E-mail address: lachkar.mohammed@gmail.com (M. Lachkar). Peer review under responsibility of King Saud University.



especially in the north of Africa and Spain (Greuter et al., 1989). Calycotome villosa subsp. intermedia has intricate, angular, pubescent branches, which are green when young and become greyish when mature. It shows sharp terminations and alternate leaves; the lower leaves are elongate, oval and trifoliate, and covered by sericeous down. The flowers are yellow and grouped, and have a bell-shaped calyx, thickly downy legume with a protuberant upper rib, and round dark olivaceous seeds. This shrub grows mostly in cool places, on asperities from 500 to 900 m above sea level (Camarda and Valsecchi, 1983). Calycotome villosa subsp. intermedia is not used in the Moroccan folk medicine. Medicinal uses have been reported only for C. villosa (Poiret) Link as antitumoral agent (Hartwell, 1982) and for the treatment of furuncle, cutaneous abscess and chilblain in the Sicilian folk medicine (Lentini et al., 1993).

http://dx.doi.org/10.1016/j.arabjc.2014.03.005

1878-5352 © 2014 King Saud University. Production and hosting by Elsevier B.V.

This is an open access article under the CC BY-NC-ND license (http://creativecommons.org/licenses/by-nc-nd/3.0/).

<sup>\*</sup> Corresponding author. Tel.: +212 671 556 742; fax: +212 535 733 171

As a part of a systematic research study on the constituents of Moroccan natural plants, we have reported, in previous phytochemical studies of Calvcotome villosa subsp. intermedia, the isolation and characterization of flavone glucosides from the leaves and flowers (El Antri et al., 2004a) and tetrahydroisoquinoline alkaloids and isoquinoline-N-oxide alkaloid from the seeds (El Antri et al., 2004b,4c). Flavonoid compounds exhibit varied biological effects and have been identified as antitumor agents, antioxidants, and free radical scavengers (Manthy and Buslig, 1998; Bors et al., 1990). Substituted-tetrahydroisoquinoline alkaloids, in which C-1 is a quaternary stereogenic center, have been reported to display very interesting biological and pharmacological properties (He et al., 2000; Herbert, 1985). In a continuing search on the seeds of this species, the methanolic extract was found to contain a complex mixture of compounds. The present study describes the isolation and the structural elucidation of a new dihydroisoquinoline-N-oxide alkaloid together with a paraben derivative.

#### 2. Materials and methods

#### 2.1. Plant material

Seeds of *Calycotome villosa* subsp. *intermedia* were collected from the aerial part of the plant in June 2011 and again in June 2012 from area of Sefrou, Morocco. The plant was identified by Dr. Abdeslam Ennabili (Sidi Mohamed Ben Abdellah University, Fez, Morocco). A voucher specimen (n° EN008) has been deposited at the herbarium of the "Institut National des Plantes Médicinales et Aromatiques", Sidi Mohamed Ben Abdellah University, Fez, Morocco.

#### 2.2. General procedures

Melting points were measured in open capillary tubes in a Büchi 530 apparatus and are uncorrected. The microanalyses have been performed at the "UATRS (Unités d'Appui Technique à la Recherche Scientifique), CNRST, Rabat, Maroc". UV-Visible spectra were obtained on a double-beam UV-Visible Camspec M550 spectrophotometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded in CDCl<sub>3</sub> on a Bruker (Wiessembourg, France) AM 300 spectrometer (300 and 75 MHz, for <sup>1</sup>H and <sup>13</sup>C NMR respectively). Chemical shifts are given as  $\delta$  value with TMS as an internal standard and coupling constants are given in Hz. ESI-MS data were obtained on a Quattro II tandem quadripole mass spectrometer (Micromass, Manchester, UK) fitted with an electrospray ionization. EI-MS data were obtained on Micromass Autospec mass spectrometer. Silica gel GF254 was used for TLC. Spots on chromatograms were detected under UV light (254 nm) and by Dragendorff's reagent. Column chromatography (CC) was carried out on silica gel 60 (70–230 mesh).

#### 2.3. Extraction and isolation

The air dried plant material (150 g) was first extracted with hexane for 24 h and then with methanol for 48 h using a Soxhlet apparatus. The methanolic solution was evaporated to dryness and the resulting crude extract was dissolved in a 5% hydrochloric acid solution, and extracted with hexane and then with  $CH_2$ .  $Cl_2$ . The aqueous solution was basified to pH 10 with concentrated ammonia and extracted three times each with 200 ml of CH<sub>2</sub>Cl<sub>2</sub>. The collected organic phases were dried over anhydrous sodium sulfate and evaporated under reduced pressure to give 2.6 g of the crude alkaloid extract. This crud alkaloid extract was subjected to column chromatography over silica gel  $(3 \times 50 \text{ cm})$  using CH<sub>2</sub>Cl<sub>2</sub>/MeOH gradient to give 15 mg of compound **2** (100% CH<sub>2</sub>Cl<sub>2</sub>) and 45 mg of compound **1** (CH<sub>2</sub>Cl<sub>2</sub>/ MeOH: 97/3). Compound **1** was re-crystallized from methanol to afford crystals suitable for X-ray analysis.

#### 2.4. Characterization data

#### 2.4.1. Compound 1

1-Hydroxymethyl-6,7-dimethoxy-3,4-dihydroisoquinoline Noxide 1, colorless powder; Mp 188–190 °C; UV  $\lambda_{max}$  nm (MeOH): 203, 240, 331; IR bands (KBr)  $v_{max}$  cm<sup>-1</sup> 3459-3349, 2970, 1610, 1520, 1465, 1352, 1274, 1186, 1142, 1029, 952 and 850; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): 3.01 (2H, t, J = 7.5 Hz, H-4); 3.82 (3H, s, 6-OCH<sub>3</sub>); 3.83 (3H, s, 7-OCH<sub>3</sub>); 4.01 (2H, t, J = 7.5 Hz, H-3); 4.73 (2H, s, 1-CH<sub>2</sub>); 6.66 (1H, s, H-5); 6.83 (1H, s, H-8); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 150.1(C-1); 148.3 (C-7); 144.6 (C-6); 124.9 (C-10); 120.5 (C-9); 110.9 (C-5); 107.6 (C-8); 57.75 (C-3); 57.7 (CH<sub>2</sub>); 56.2 (CH<sub>3</sub>-O); 56.1 (CH<sub>3</sub>-O); 27.5 (C-4); EA: Found: C, 60.72; H, 6.39; N, 5.90. C<sub>12</sub>H<sub>15</sub>NO<sub>4</sub> requires: C, 60.75; H, 6.37; N, 5.91; O, 26.97%. HRMS ESI (*m*/*z* found 238.2591, calc.: 238.1001 for C<sub>12</sub>H<sub>15</sub>NO<sub>4</sub>); ESI-MS *m*/*z*: 238 [M+H]<sup>+</sup> (C<sub>12</sub>H<sub>15</sub>NO<sub>4</sub>), 220, 190, 175.

#### 2.4.2. Compound 2

Methyl-4-hydroxybenzoate **2**, colorless needles; Mp 126–128 °C; UV  $\lambda_{max}$  nm (MeOH): 288, 325; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): 7.95 (2H, d, J = 8.7 Hz, H-3, H-5); 6.84 (2H, d, J = 8.7 Hz, H-2, H-6); 5.57 (1H, s, OH); 3.86 (3H, s, OCH<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 167.0 (C-7); 159.8 (C-4); 131.9 (C-2 and C-6); 122.8 (C-1); 115.2 (C-3 and C-5); 52.0 (CH<sub>3</sub>-O). HRMS ESI (m/z found 152.0445, calc.: 152.0473 for C<sub>8</sub>H<sub>8</sub>O<sub>3</sub>); EI-MS m/z: 152 [M]<sup>+</sup> C<sub>8</sub>H<sub>8</sub>O<sub>3</sub>, 121, 93, 65.

#### 2.5. X-ray crystal structure determination

Colorless single crystals of 1, suitable for X-ray structure analysis, were obtained by slow evaporation of a methanol solution. The crystal has been collected on a STOE IPDS II two-circle-diffractometer using a MoK<sub> $\alpha$ </sub> radiation ( $\lambda = 0.71073$  Å). For further information on crystal structure and refinements data, CCDC 225920 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/ retrieving.html (or from the CCDC, 12 Union Road Cambridge CB2 1EZ, UK; fax: +44 1223 336033; e-mail: deposit@ccdc.cam.ac.uk).

#### 3. Results and discussion

The seeds of *Calycotome villosa* subsp. *intermedia* were extracted with hexane and then with hot methanol. The methanolic extract was purified in the usual way for the isolation of alkaloids (Dalton, 1979). Fractionation and silica gel column chromatography yielded the new compound **1** and the



Scheme 1 A = phenyl ring and B = sofa six-membered ring.

known compound **2** (Scheme 1) besides the previously isolated alkaloids (El Antri et al., 2004b,4c). Compound **1** was the less polar of all alkaloids found in the methanolic extract of the seeds of *Calycotome villosa* subsp. *intermedia*.

Compound 1 was isolated in a crystalline form and completely characterized. The molecular formula of 1 was determined to be  $C_{12}H_{15}NO_4$  on the basis of ESI-MS (m/z = 238 $[M+H]^+$ ) and elemental analysis. In the UV spectrum of 1, the maximum bands are 203, 240 and 331 nm. Its IR spectrum showed absorption bands at 3459-3349, 2970, 1610, 1520, 1465, 1274, 1142, 1029 and 850 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectrum of 1 showed two aromatic protons at  $\delta$  6.83 (s, H-8) and 6.66 (s, H-5), two signals at  $\delta$  4.01 (t, 2H, J = 7.5 Hz) and 3.01 (t, 2H, J = 7.5 Hz) typical of H-4 and H-3 protons of a dihydroisoquinoline nucleus, and two singlets at  $\delta$  3.83 and 3.82 corresponding to the methoxyl groups at C-6 and C-7, which, in addition to the <sup>13</sup>C NMR spectrum and a prominent peak at m/z = 238 in the ESI mass spectrum, confirmed the dihydroisoquinoline skeleton in the molecule. The remaining <sup>1</sup>H and <sup>13</sup>C NMR data of 1 (singlet at  $\delta$  4.73 (2H), three CH<sub>2</sub>, two CH and five quaternary carbons) as well as the presence of an ion at  $m/z = 190 [C_{12}H_{15}NO_4 + H - H_2O - NO]^+$  in the ESI-MS, evidenced a dihydroisoquinoline N-oxide in the molecule. Thus, the structure of 1 was determined as a 1hydroxymethyl-6,7-dimethoxy-3,4-dihydroisoquinoline-Noxide, which is a new natural product.

Compound **2** was isolated as colorless needles, with molecular formula  $C_8H_8O_3$  deduced from the molecular ion peak at m/z 152 in the EI-MS and supported by <sup>1</sup>H NMR and <sup>13</sup>C NMR data. The mass spectrum of compound **2** showed molecular ion peak at m/z 152 with ions corresponding to loss of OCH<sub>3</sub> (M - 31; m/z = 121) followed by loss of the carbonyl group (M - 31 - 28; m/z = 93). The <sup>1</sup>H NMR spectrum analysis re-

Table 2	Selected bond leng	gths and angles of 1.		
Bond lengths [Å]				
C1 C2	1.383(2)	C5 C6	1.398(2)	
C1 C6	1.411(2)	C5 C10	1.506(2)	
C2 C3	1.412(2)	C7 N8	1.311(2)	
C3 C4	1.386(2)	C9 C10	1.517(2)	
Bond angles [°]				
C1 C2 C3	119.8(1)	C7 N8 C9	121.9(1)	
C3 C4 C5	120.8(1)	N8 C7 C6	119.6(1)	
C5 C6 C1	119.2(1)	C5 C10 C9	111.5(1)	
C5 C6 C7	120.3(1)	N8 C9 C10	111.1(1)	

vealed the presence of a pair of doublets at  $\delta$  7.96 (d, J = 8.2 Hz, H-3 and H-5) and 6.88 (d, J = 8.2 Hz, H-2 and H-6), each integrating to two protons, suggesting the presence of two aromatic hydrogen in ortho position. Each singlet of hydroxyl and methoxyl signals was observed at  $\delta$  5.60 and 3.92, respectively. Its <sup>13</sup>C NMR spectrum showed six resonances (two quaternary carbons, two CH groups, one carboxyl and one methoxyl) and the signals of carboxyl C=O ( $\delta$  167.0) and methoxyl ( $\delta$  159.8). Accordingly, compound 2 was assigned as a paraben derivative named methyl-4-hydroxybenzoate (Scheme 1). The physical and spectroscopic data found are in concordance with literature values (Cheng et al., 2001; Perveen et al., 2009). Although there are many types of paraben derivatives by organic synthesis (Crombie et al., 1979; Hirota et al., 1981), methyl-4-hydroxybenzoate 2 was isolated for the first time from Calvcotome villosa subsp. intermedia.

The crystal structure of **1** has been determined on single crystal. Crystallographic data are given in Table 1. Selected geometrical parameters, bond lengths and angles are listed in Table 2 according to the numbering scheme displayed in Fig. 1. The unit cell contains two molecules  $C_{12}H_{15}NO_4$ . The main part of the molecule of alkaloid **1** is the dihydroisoquinoline molecular fragment, which consists of the phenyl ring (A) and the sofa six-membered ring (B) (see Scheme 1). The deformation of the molecule can be expressed in terms of the distorsion angles between its different parts. According to the values of the corresponding distorsion angles: C3 C4 C5 C10–175.5(2), C7 C6 C5 C4–176.7(2), C10 C5 C6 C1 176.3(2), C2 C1 C6 C7 177.1(2), C1 C2 C3 O31–178.4(2), we

Table 1 Crystal data and measurement conditions of compound 1.			
Empirical formula	$C_{12}H_{15}NO_4$		
Formula weight (g/mol)	237.25		
Temperature (K)	100(2)		
Wavelength $[\lambda K_{\alpha} (A)]$	0.71073		
Crystal system	Triclinique		
Space group	P-1		
Unit cell dimensions (Å, °)	a = 6.9573 (12)	$\alpha = 74.987(12)$	
	b = 8.6906 (13)	$\beta = 87.066 \ (14)$	
	c = 9.7297 (16)	$\gamma = 77.254$ (13)	
Volume ( $Å^3$ )	554.20 (16)		
Z	2		
$d_{\rm cal.}~({\rm g/cm}^3)$	1.422		
Absorption coefficient (mm <sup>-1</sup> )	0.107		
F(000)	252		
Crystal size (mm <sup>3</sup> )	$0.36 \times 0.32 \times 0.22$		
Reflections collected	6426		
Final R indices $[I > 2\sigma (I)]$	R1 = 0.0348, wR2 = 0.0802		





Figure 1 Perspective view of the molecule of 1. Thermal ellipsoids drawn at 50% probability levels.



**Figure 2** Projection of the crystal structure of alkaloid 1 onto [001].

state that the bonds C5–C10 and C6–C7 are in the plane of the ring. Same could be concluded about the orientation of the methoxy groups. The O31–C32 methoxy group lies in the plane of the ring (C4 C3 O31 C32 is  $-4.2^{\circ}$ ) while the methoxy group O21 C22 is out of the plane (C1 C2 O21 C22 is  $16.7^{\circ}$ ). The hydroxyl group O72 has a gauche arrangement with respect to the H72–O72–C71–C7 torsion angle ( $65.0^{\circ}$ ).

A one restriction to the freedom of the molecule's motion is imposed by intramolecular H-bonds, between H72 and O81 (2.711(2)Å). More such interactions are used to extend the molecular network in the three dimensions. Fig. 2 depicts a projection on the [001]. One can remark the formation of the "dimers" and their propagation in the ab-plane using H–O interactions.

#### 4. Conclusion

The isolation and spectroscopic characterization of a new alkaloid together with a paraben derivative from the seeds of *Calycotome villosa* subsp. *intermedia* was described. The structure of the new alkaloid **1** was solved by X-ray analysis. Paraben derivative **2** was isolated for the first time from *Calycotome villosa* subsp. *intermedia*.

#### Acknowledgements

This work was funded by grants from the « Centre National de la Recherche Scientifique et Technique, Rabat, (URAC 19) » and the University Sidi Mohamed Ben Abdellah (Fez). The authors thank Dr. Abdeslam Ennabili (Sidi Mohamed Ben Abdellah University, Fez, Morocco) for identification of the plant.

#### References

- Bors, W., Heller, W., Michel, C., Saran, M., 1990. Flavonoids as antioxidants: determination of radical scavenging efficiencies. Methods Enzymol. 186, 343–355.
- Camarda, I., Valsecchi, F., 1983. Alberi e arbusti spontanei della Sardegna. In: Gallizzi, G. (Ed.). Sassari, Italy, pp. 241–242.
- Cheng, M.J., Tsai, I.L., Chen, S., 2001. Chemical constituents from *Strychnos cathayensis*. J. Chin. Chem. Soc. 48, 235–239.
- Crombie, L., Eskins, M., Games, D.E., Loader, C., 1979. Polyketoenols and chelates. Synthesis of acetyl- and alkoxycarbonylxanthyrones. J. Chem. Soc. Perkin Trans. 1, 478–482.
- Dalton, D.R., 1979. The Alkaloids The Fundamental Chemistry, a Biogenetic Approach. Marcell Dekker, New York.
- El antri, A., Messouri, I., Chendid Tlemcani, R., Bouktaib, M., El alami, R., El bali, B., Lachkar, M., 2004a. Flavone glycosides from *Calycotome villosa* subsp. *intermedia*. Molecules 9, 568–573.
- El Antri, A., Messouri, I., Bouktaib, M., El Alami, R., Bolte, M., El Bali, B., Lachkar, M., 2004b. Isolation and X-ray crystal structure of a new isoquinoline-N-oxide alkaloid from *Calycotome villosa* subsp. *intermedia*. Fitoterapia 75, 774–778.
- El Antri, A., Messouri, I., Bouktaib, M., El Alami, R., Bolte, M., El Bali, B., Lachkar, M., 2004c. Isolation and X-ray crystal structure of tetrahydroisoquinoline alkaloids from *Calycotome villosa* subsp. *intermedia*. Molecules 9, 650–657.
- Gibbs, P.E., 1968. Taxonomy and distribution of the genus Calicotome. Notes Roy. Bot. Gard. 28, 275–286.
- Greuter, W., Burdet, H.M., Long, G., 1989. A critical inventory of vascular plants of the circum-mediterranean countries. Dicotyledones (Lauraceae-Rhamnaceae). Geneva, Switzerland, Med-Checklist 4.
- Hartwell, J.L., 1982. Plants Used Against Cancer. A survey, USA7 Quarterman Publications, Lawrence, MA.
- He, Y., Nikulin, V.I., Vansal, S.S., Feller, D.R., Miller, D.D., 2000. Synthesis and human beta-adrenoceptor activity of 1-(3,5-diiodo-4methoxybenzyl)-1,2,3,4-tetrahydroisoquinolin-6-ol derivatives in vitro. J. Med. Chem. 43, 591–598.
- Herbert, R.B., 1985. In: Philipson, J.D., Roberts, M.F., Zenk, M.H., (Eds.), The Chemistry and Biology of Isoquinoline Alkaloids, Springer Verlag: Berlin, Heidelberg, New York, Tokyo, p. 213.
- Hirota, K., Kitade, Y., Senda, S., 1981. Pyrimidine deriva-tives and related compounds. 39. Novel cycloaromatiza-tion reaction of 5formyl-1,3-dimethyluracil with three-carbon nucleophiles. Synthesis of substituted 4-hydroxy-benzoates. J. Org. Chem. 46, 3949–3953.
- Lentini, F., Aleo, M., Amenta, R., 1993. L'uso popolare delle piante nelle Isole Egadi (Sicilia). Giorn. Bot. Ital. 127 (3), 702.
- Manthy, J.A., Buslig, B.S., 1998. Flavonoids in the Living System; in Advances in Experimental Medicine and Biology. Plenum, New York.
- Perveen, S., Malik, A., Bakhsh Tareen, R., 2009. Phytochemical studies on *Perovskia atriplicifolia*. J. Chem. Soc. Pak. 31 (2), 314–318.
- Tutin, G.T., 1972. Flora Europea. University Press, Cambridge.